

1 **PHYTOCHEMISTRY AND BIOLOGICAL ACTIVITIES OF LEAVES AND PULP EXTRACTS FROM**
2 **ZIZYPHUS MAURITIANA (LAM.) COLLECTED IN MALI**

3
4 **ABSTRACT:**

5 **Introduction:** The objectives of this work were to make a phytochemical study of the leaves and pulp
6 of *Zizyphus mauritiana* Lam collected from the flooded and exonded sites of Niono and Sévaré in Mali
7 on the one hand and to evaluate the biological activity of the extracts notably the antiradical activity
8 and hemolytic activity extracts on the other hand.

9 **Method:** Phytochemical analysis and biological activities were performed at the plant biochemistry
10 and biotechnology laboratory of the FST / USTTB. The tannins and flavonoids were extracted by
11 maceration and assayed by spectrophotometry and the saponosides were extracted by decoction.

12 **Results:** Catechin tannins, flavonoids and abundant sugars in the leaves and pulp and saponosides
13 are absent in the pulp of our samples.

14 Calcium, carbonate and chloride ions are abundant in the leaves and pulps of the excavated site of
15 Niono and the flooded Sévaré site.

16 HPLC chromatograms of leaf tannin and pulp extracts showed two peaks of gallic acid.

17 The antiradical activity on the DPPH of the leaf extracts would be linked to the collection site Catechin
18 tannins and flavonoids of the flooded sites have a greater antiradical activity than those of the
19 exonded sites of Niono and Sévaré, this is not the case extracts of pulp. Their antiradical activity
20 remains lower than that of ascorbic acid whose IC₅₀ is 30 µg.

21 Leaf saponosides showed good hemolytic activity on red blood cells, especially those from the
22 flooded Niono site and the Sévaré extruded site.

23 **Conclusion:** Secondary metabolites such as phenolic compounds have good antiradical activity and
24 saponosides extracted from the leaves have hemolytic activity.

25 **Key words:** Phenolic compounds, tannins, flavonoids, saponosides, *Zizyphus mauritiana* Lam,
26 biological activity.

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Keywords are usually arranged in alphabetical

27 **1. INTRODUCTION**

28 *Zizyphus mauritiana* (Lam) has many nutritional, medical, artisanal and even orchard protection
29 interests. Several previous works have shown its richness in primary and secondary metabolites [1],
30 [2], [3] as well as its economic interests [4]. Danthu et al. Studied phytochemical composition in two
31 wild and domestic species in Senegal [3].

32 These metabolites have antibacterial, analgesic, astringent and anti-inflammatory properties, which
33 can justify their use in traditional medicine [5], [6]. The fruits of the plant are an important source of
34 income for many rural families [4].

35 Other studies have shown the hypoglycemic, hypertensive, anti-inflammatory, antibacterial and
36 antioxidant activity of leaf extracts [7], [8], [9]. The antioxidant power of tannins (catechical and gallic
37 tannins) and flavonoids would help fight against aging [7].

38 A comparative study of certain abundant metabolites in the leaves and pulp of the plant would make it
39 possible to evaluate the nutritional and biological qualities according to the collection sites (site
40 flooded, site exonde). During these works metabolites were extracted characterize and assay. Their
41 anti-radical activities on DPPH and hemolytic on red blood cells of beef blood were evaluated

42 **2. MATERIAL AND METHOD::**

43 **2.1. Equipment :**

44 **2.1.1. Sample collection sites:**

45 Samplings of leaves and pulps of *Zizyphus mauritiana* (Lam.) Were collected at the following sites: In
46 Niono flooded site: Sitan Wéré, and exonded site: Ranch. In Sévaré flooded site Dialagou and site
47 exonded: Doundoun.

48 **2.1.2. Plant material:**

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next page so that it is not separate from its content

49 Biological material consisted of powder, leaves and pulp of *Zizyphus mauritiana* (Lam.) Sites of Niono
50 and Sévaré, the Blood of beef has been taken at the slaughterhouse of Sabalibougou in Kati.

51 2.2. METHODS :

52 2.2.1. Characterization reactions of metabolites:

53 2.2.1.1. Catechetal tannins:

54 The catechin tannins were extracted by maceration in 100 mg of plant organ powder in acetone
55 diluted to 7/3 for 20 minutes with stirring, the filtrate was concentrated in a rotavapor saturated with
56 sodium chloride and centrifuged at 3000rpm. for 10 minutes. The catechical tannins have been
57 characterized by ferric chloride [7]

Comment [U3]: what is the mesh size of powder particles?

58 2.2.1.2. Gallic tannins:

59 Gallic tanins were extracted in one hundred milligram (100 mg) of organ powder delipidated with
60 petroleum ether and then boiled in 20 mL of distilled water for 10 minutes. The dichloromethane is
61 mixed with the filtrate to remove the pigments. Gallic tannins were extracted in the aqueous phase
62 with ethyl acetate and evidenced by 2% ferric chloride [10], [7].

63 2.2.1.3. Flavonoids:

64 In a test tube ten (10) drops of concentrated hydrochloric acid was added to 0.5mL of extract and a
65 few milligrams of magnesium turnings. After three minutes of incubation at room temperature, specific
66 staining of flavonoids was observed [11], [12], [7].

Comment [U4]: the nominal value must be written how many milligrams

67 2.2.1.4. Coumarines:

68 Five milliliters (5 mL) of etheric extract (maceration for 24 hours) were evaporated in a beaker in the
69 open and then 2 mL of hot water was added to the residue. The solution was partitioned between two
70 tubes and 0.5 mL of 25% NH₄OH was added to the contents of one of the tubes and then mixed.
71 Fluorescence was observed at UV at 366 nm [9], [10]. **What is the other tube for? used as a
72 reference ?**

73 2.2.1.5. Leucoanthocyanes:

74 To 5 ml of infused prepared from the drug powder, 5 ml of sulfuric acid and then 5 ml of NH₄OH were
75 added to a test tube and the appearance of leucoanthocyanin-specific staining was followed [10] .

76 2.2.1.6. Sterols and triterpenes:

77 In a test tube, one (1) gram of organ powder was added twenty (20) mL of petroleum ether. The
78 solution was stirred and left in the refrigerator for 24 hours, then filtered on filter paper in a beaker and
79 evaporated to dryness in a rotavapor. The sterols and triterpenes were extracted in the residues with
80 10 mL of chloroform. To 10 mL of chloroformic extract we added 1 mL of acetic anhydride and 1 mL
81 of CHCl₃. The chloroform solution was split into two test tubes, at the bottom of one of the tubes 2 mL
82 of concentrated H₂SO₄ was deposited and the other tube was used as a reference. The tube should
83 not be shaken and the appearance of specific staining was followed.

84 2.2.1.7. Mucilages:

85 The mucilages were extracted by maceration of five grams of plant organ powder (leaves and pulp) in
86 one hundred (100) mL of distilled water for 12 hours. Mucilages were precipitated by ethanol.
87 Obtaining a fluffy precipitate by mixing indicates the presence of mucilage [9], [10], [11].

88 2.2.1.8. Reducing sugars:

89 The reducing sugars have been characterized by hot Fehling liquor. For this purpose, 1 milliliter of
90 Liquor and Organ Extract Reagent was mixed to volume and then boiled at 80 ° C for 5 minutes.

Comment [U5]: What does it mean ?

91 2.2.2. Characterization of some mineral salts of leaves and pulps:

92 One gram of the organ powder was calcined in an oven at 600 ° C for 12 hours. The ash obtained is
93 weighed and then dissolved in 10 ml of distilled water and filtered. The different ions were highlighted
94 in the filtrate.

95 **Table I:** Characterization Reaction of Mineral Salts

Comment [U6]: We recommend that tables on one page. The letters in the table must be smaller than the text.

Ions	Reagents	Results
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Phosphate	Hot ammonium nitro-molybdate	Yellow precipitate
Sulfate	Barium chloride	White precipitate
Calcium	Ammonium oxalate	White precipitate
carbonates	Acid chlorihydric on ash	Effervescence reaction
Chloride	Silver nitrate	White precipitate, darkens in the light
Potassium	Cobalt sodium nitrit	Needle crystal

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2.2.3. Dosage of catechical and gallic tannins, flavonoids and saponosides:

2.2.3.1. Catechetical tannins:

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The content of the catechical tannins in the extracts was determined spectrophotometrical. In a test tube were mixed (1 mL) of acetone extract, 5 mL of distilled water, 1 mL of ethanol and 0.5 mL of Folin reagent. After standing for 5 minutes, 1 ml of 5% sodium carbonate solution was added and left in the dark for 1 hour. Absorbance reading was made at 725 nm. A 1% gallic acid standard range of 10 to 100 µg was used [9].

2.2.3.2. Flavonoids:

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To 500 µL of the extract were added 2 mL of distilled water and then 150 µL of 5% sodium nitrite (NaNO₂). After five minutes, 100 (100 µL) of aluminum trichloride (AlCl₃) at 10% (w / v) is added to the mixture. After six (6) minutes of incubation at room temperature 1 ml of 1M sodium carbonate (NaCO₃) is added. The content was homogenized and the absorbance of the pinkish solution was determined at 510 nm against a blank. Catechin was used as a positive control. The total flavonoid content of plant extracts is expressed in milligram (mg) equivalent of catechical per 100 grams of dry vegetable matter (mg EC / 100g) [9], [11].

2.2.3.3. Saponosides:

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Dosage of the saponosides was done by calculation of the foam index.

The extraction was done as follows. To 1 g of organ powder, 5 ml of petroleum ether was added to delipidate for 5 minutes. The supernatant is poured and the operation is repeated with 2.5 mL of petroleum ether. The powder was dried at laboratory temperature. To 0.5 g of delipidated powder was added 10 mL of distilled water and the mixture was boiled with stirring for twenty (20) minutes and filtered. The decoction was used to determine saponoside content by 1/10 dilution [10].

Each tube was shaken horizontally for 15 seconds and allowed to stand for 15 minutes. The foam index was calculated in the tube having 1 cm of foam height. That is a 1/10 dilution of the 1% decoction at a concentration of 0.1%.

If the tube containing 5 mL of decoction and 5 mL of distilled water has a foam height of 1 cm, the 5 mL of 1% has 0.05 g of drug and the foam number is $10 * 1 / 0.05 = 200$.

2.2.3.4. Liquid chromatography of tannin extracts:

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The standards were prepared in a 50/50 (v / v) water / methanol mixture. Several calibration ranges were used: 10 mg / ml, 20 mg / ml, 50 mg / ml to establish the calibration line.

Lyophilized tannic extracts were dissolved in the 50/50 (v / v) water / ethanol mixture. Then they were sonicated for 15min, allowed to cool to room temperature and filtered through a nylon membrane with 0.45µm pores prior to injection.[18]

LC conditions:

Mobile Phase: Water / 20mM Phosphate / Acetonitrile buffer 70: 28: 2 v / v / v

Column: C18, 4.6 x 150mm, 5µ- Zorbax- Agilent

Flow rate: 0.8ml / min, injection volume: 20µl, column temperature: 30 ° C, detection: 271nm

2.2.4. Antioxidant activity: 1-1 diphenyl-2-pyrryl hydrazyl test (DPPH):

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The antioxidant activity of the aqueous extracts of *Z. mauritiana* L. and of a standard antioxidant (ascorbic acid) with respect to the DPPH radical was evaluated using a spectrophotometer by following the reduction of this radical which is accompanied by its passage from the purple color

149 (DPPH) to the yellow color (DPPH). A negative control tube is prepared by replacing the extract with
 150 distilled water. The tubes were placed in the dark for 30min and the reading was made at 517nm. [12]
 151 The results are expressed in% of anti-radical activity or Inhibitory in percentage (I%) according to the
 152 formula: I% = [(Abs negative control-Abs Sample) / Abs control [13],
 153 The IC₅₀ of each extract was calculated from a linear regression line established with the percentages
 154 of inhibition obtained. IC₅₀ is the concentration of the extract that inhibits 50% of the activity of the
 155 radical, plus it is small plus the extract is considered a powerful antioxidant [14].
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157 2.2.5. Hemolytic activity of saponosides:

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 159 The tests were performed on red blood cell pellets obtained by centrifugation at 4000 rpm for five (5).
 160 They were washed three (3) times with physiological saline buffered because of 1ml of blood
 161 dissolved in 25ml of saline.

162 In a series of six tubes each containing 0.5 ml of packed red blood cells were added increasing
 163 volumes of a solution at 1 mg / ml or 5 mg / ml of saponosides extracted from the leaves and 2 ml of
 164 the buffered saline solution. The mixture was homogenized and the tubes were allowed to stand for
 165 24 hours and then centrifuged at 3500 rpm for 10 minutes. [16]

166 The turbidity through the red coloration observed in the tubes containing the extract and the pellet
 167 determines the haemolytic activity. The absorbance of the solutions of the tubes

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169 3. RESULTS AND DISCUSSION:

170 3.1.Characterization of metabolites:

171 The results obtained after the characterization reactions of the metabolites are recorded in the table
 172 below.

173 Table II: Characterization of primary and secondary metabolites

Metabolites	Feuilles				Pulpes			
	Niono		Sévaré		Niono		Sévaré	
	exonded site	Flooded site	exonded site	Flooded site	exonded site	Flooded site	exonde d site	Flooded site
Catechical tannins	+++	+++	+++	+++	+++	+++	+++	+++
Gallic tannins	++	++	+	+	++	++	+	+
flavonoids	+++	+++	+++	+++	++	++	++	++
coumarins	+	++	+	++	+	+	++	+
leucoanthocya nins	++	+	+	+++	+	+	+	++
saponosides	+++	++	++	+++	-	-	-	-
Terpenes - Sterols	++	++	++	++	+	++	+	++
mucilage	++	++	++	++	++	++	++	++
sugars	+++	+++	+++	+++	+++	+++	+++	+++

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 175 Legend: +++ = Very abundant, ++ = Not abundant, + = Traces, - = Absent

176 Table III: Characterization of mineral salts in leaves and pulps

	Leaves				Pulps			
	Niono		Sévaré		Niono		Sévaré	
	Exonded Site	Flooded site	Exonded Site	Flooded site	Exonded Site	Flooded site	Exonded Site	Flooded site
Sulfate	+	+	+	+	+	+	+	+
Calcium	++	++	++	+++	+++	+++	++	++
Carbonate	++	++	++	+	+	+	++	++
Chlorure	++	++	++	+	+	+	++	++
Potassium	+	+	+	++	++	++	+	+

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 178 Legend: +++ = Very abundant, ++ = Not abundant, + = Traces, - = absent

179 3.2. Determination of metabolite of leaves and pulp:

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180 Table IV and table V shows the results of catechical tannin determination at the leaves and pulps of
 181 the different sites. These results are the averages of three trials.

182 Table IV: Content of extracts in leaves

Samples	Percentage			
	Sites de Niono		Sites de Sévaré	
	Exonded Site	Flooded site	Exonded Site	Flooded site
Content in catechical tannins	2,40 ± 0,04 ^a	2,88 ± 0,03 ^b	2,03 ± 0,03 ^a	3,13 ± 0,03^a
Content in gallic tannins,	2,32 ± 0,04	0,60 ± 0,02	0,52 ± 0,03	1,07 ± 0,03
Flavonoids content	1,11 ± 0,3	0,83 ± 0,2	0,35 ± 0,3	1,17 ± 0,2

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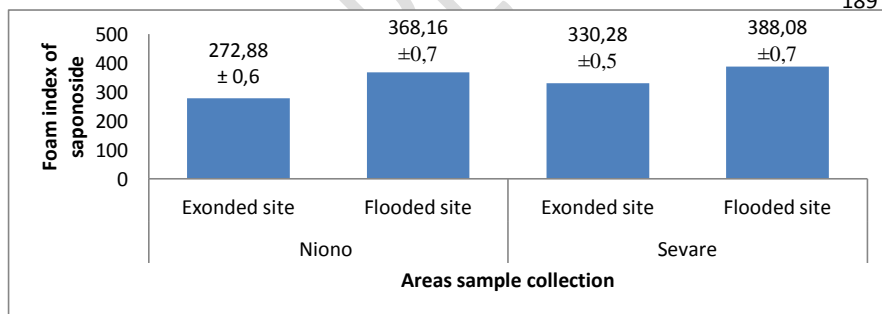
184 Table V: Content of extracts in pulp

Samples	Content			
	Niono		Sevare	
	Exonded Site	Flooded site	Exonded site	Flooded site
Content in catechical tannins	3,00 ± 0,02	3,25 ± 0,05	4,02 ± 0,05	2,39 ± 0,04
Content in gallic tannins,	2,62 ± 0,04	1,90 ± 0,04	1,19 ± 0,04	2,12 ± 0,04
Flavonoids content	1,70 ± 0,58	2,14 ± 0,5	3,55 ± 0,5	3,28 ± 0,6

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186 **3.3. Dosage of saponosides in the leaves:**

The foam index of the saponosides was calculated in the tube whose foam height is equal to 1 cm or close to the different samples of *Zizyphus mauritiana* Lam



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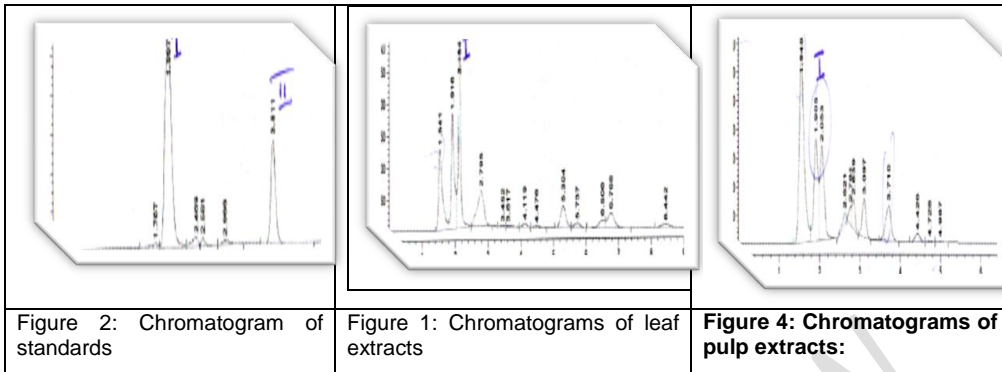
192 Figure 1: Saponoside content of leaf extracts from Niono and Sévaré sites

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194 **3.4. Qualitative analysis by HPLC of tannic extracts:**

195 The following chromatograms were obtained by HPLC extracts of the leaves and pulp of the localities
 196 of Niono and Sévaré.

197 Gallic acid = (I): Retention time = 1.997min and Catechol = II: Retention time = 3.811min

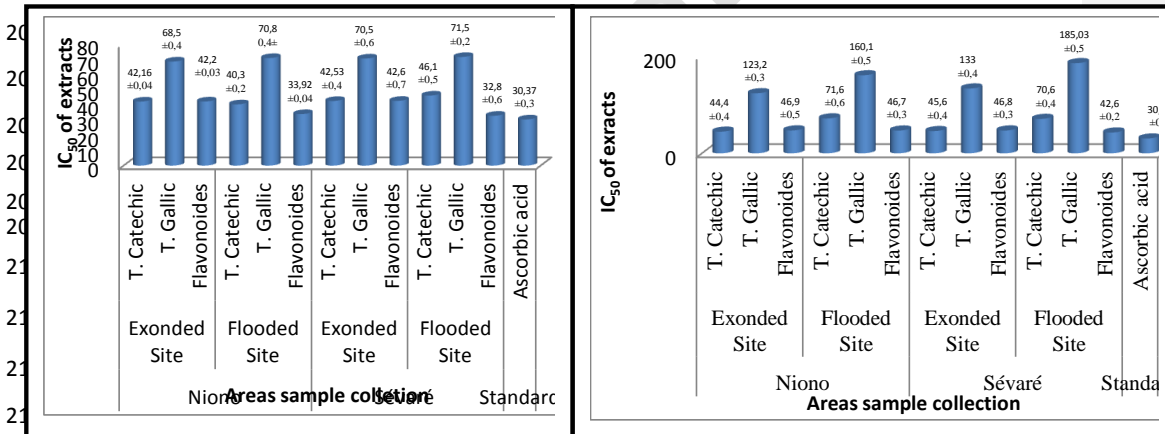


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199 **3.5. Biological activity:**

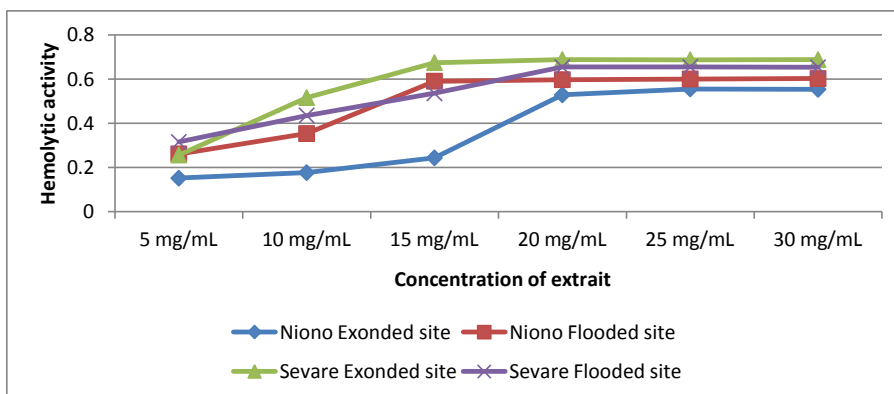
200 **3.5.1. Antioxidant activity of leaf and pulps extracts:**

201 The antiradical activity of leaf extracts from Niono and Sévaré sites was evaluated by their
 202 concentration which inhibits 50% of the radical IC_{50} from equations of the linear regression line of the
 203 percentages of inhibition (% I)



214 The optical densities of the supernatants obtained after centrifugation gave the following results

215 We recommend this sentence moved to the next page, this space for title Figure 5 and Figure 6.



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217 Figure 7: Curve of Hemolytic Activity of Saponoside Extracts of 5 mg / mL

218 Tannins, flavonoids and catechical sugars have been found abundant in the leaves and pulps of the
 219 samples, whereas The gallic tannins, coumarins, leucoanthcyanins and mucilages were less
 220 abundant. It should be noted that the saponosides are absent in the pulp of all the samples of the
 221 different sites Table II.

222 Sulphate ions, calcium ions, carbonate ions, chloride ions and potassium ions were found in the
 223 leaves and pulps of the no flooded and flooded sites of Niono and Sévaré Table III

224 The leaves of the flooded sites of Niono and Sévaré have an average rate in catechical tannins
 225 higher than those of the exposed sites of Niono. Flavonoids and gallic tannin production is not related
 226 to the types of areas. In fact, the flooded sites of Niono and flooded with Sévaré have a higher
 227 content.

228 The samples from no flooded of Niono (Ranch) and the flooded Sévaré (Dialogou) site show the
 229 highest flavonoid levels at 1.11 ± 0.3 and 1.17 ± 0.2 , respectively.

230 These results are similar to those of Souhila et al., Obtained in the bracts of *Cynara scolymus* L by
 231 maceration in water at 2.39%, in 2.15% ethanol, in acetone. 2.82% in 1.99% methanol. They found in
 232 flowers by maceration in water 3.53%, in ethanol 3.75%, in acetone 2.74% and in methanol 2.05%
 233 [11].

234 Saponosides are abundant in the leaves of the Niono and Sévaré samples and absent in the pulps.
 235 The highest foam index is that of the flooded site of Sevare (Dialogou) with 388 and non-flooded site
 236 of Niono have the lowest index with 239.

237 The flavonoids of the flooded sites have a greater antiradical activity on DPPH than those of the non-
 238 flooded sites of Niono and Sevare with IC_{50} of $33.92 \pm 0,04 \mu g$ and $32.8 \pm 0,6 \mu g$ respectivement. The
 239 catechical tannins extracted from the Niono sites with IC_{50} of $42.16 \mu g$ and $40.2 \pm 0,3 \mu g$ have a greater
 240 antiradical activity than extracts from the Sévaré sites. Gallic tannins have less activity activity
 241 antiradicalaire.

242 At the level of the pulp the flavonoids of the flooded site of Sévaré have more important activity
 243 followed catechical tannins of the exonated site of Niono with IC_{50} of $42.6 \pm 0,,2 \mu g$ and $44.4 \pm 0,04 \mu g$.
 244 The IC_{50} values are very close to those of ascorbic acid, $30.37 \pm 0,3 \mu g$. The IC_{50} values are very close
 245 to those of ascorbic acid, $30.37 \mu g$ graphe These results are close to those of Nabila who obtained
 246 90% [13] for bile tannins, the% is between 55.5% and 67.4%.and different for flavonoids whose the
 247 percentage inhibition is between 36.1% and 59.4%

248 At the level of saponoside extracts from the flooded Niono site and the exonded Sévaré site, total
 249 haemolytic activity was observed from 15 mg of saponins. Whereas at the level of the Sévaré flooded
 250 site and the non- flooded site of Niono, the total haemolytic activity was obtained from 20 mg of
 251 saponoside. Hemolytic activity would not be linked to collection sites.

252 These results are similar to those of Ouedraogo et al., Who achieved total hemolytic activity with 21
 253 mg of stem extracts and 15 mg of *Mitragyna inermis* root extract [17]. Najiba obtained a 54.21%
 254 haemolytic activity with total alkaloid extracts at 5 mg / ml of *Berberis Vulgaris* L. [13]. Our results are
 255 different from those of Haddouchi et al., Whose haemolysis test showed that four species had a weak
 256 haemolytic effect [18]

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258 **4. CONCLUSION:**

259 Tannins, flavonoids and saponosides were extracted characterized and assayed in the samples
260 collected on the exonated and flooded sites of the Niono and Sévaré sites. . Their biological activities
261 have been assessed through

262 In conclusion, leaf and pulp samples of *Zizyphus mauritiana* Lam from the flooded and exposed sites
263 of Niono and Sévaré are rich in mineral salts such as phosphates, sulphates, calcium, carbonates and
264 potassium. Their production in the leaves or pulp is not related to the collection site. Secondary
265 metabolites have been found in the leaves as well as in the pulp such as catechic and gallic tannins,
266 flavonoids, coumarins, leucoanthocyanins, terpenes and sterols, mucilages and saponosides.
267 Catechin tannins are more abundant in the leaves than gallic tannins.

268 We note that extracts of catechic tannins and flavonoids from leaves and pulps showed good anti-
269 radical activity on DPPH. This activity would be linked to the collection site at the leaf level extracts
270 from flooded sites of Niono and Sévaré showed a greater antiradical activity than extracts from the
271 excavated sites. At the pulp level, the antiradical activity would not be linked to the site. The three
272 collection sites have almost the same IC50 except that of the flooded site of Sévaré, which is inferior
273 to the others and therefore more active. Their antiradical activity remains lower than that of ascorbic
274 acid.

275 Leaf saponosides showed good haemolytic activity on red blood cells, especially those from the
276 flooded Niono site and the Sévaré extruded site. Hemolytic activity would not be related to the
277 collection site.

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280 **Competing interests::**

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282 Authors have declared that no competing interests exist.

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284 **Références Bibliographies:**

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